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				spectra				
NEWS	16	MAR	31	CA/CAplus and CASREACT patent number format for U.S.				
				applications updated				
NEWS	17	MAR	31	LPCI now available as a replacement to LDPCI				
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NEWS	19	APR	04	STN AnaVist, Version 1, to be discontinued				
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				predefined hit display formats				
NEWS	21	APR	28	EMBASE Controlled Term thesaurus enhanced				
NEWS	22	APR	28	IMSRESEARCH reloaded with enhancements				
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				sequence search option				
NEWS		JUN	06	EPFULL enhanced with 260,000 English abstracts				
NEWS		JUN		KOREAPAT updated with 41,000 documents				
NEWS	27	JUN	13	USPATFULL and USPAT2 updated with 11-character				
				patent numbers for U.S. applications				
NEWS	28	JUN	19	CAS REGISTRY includes selected substances from				
				web-based collections				

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=> s crown ether (P2) (amino acid or peptide or protein or polypeptide)
MISSING OPERATOR 'ETHER (P2'

The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

=> s (crown (w) ether) (P2) (amino acid or peptide or protein or polypeptide) MISSING OPERATOR ETHER) (P2

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The search profile that was entered contains terms or
nested terms that are not separated by a logical operator.
=> s (crown (w) ether) (P) (amino acid or peptide or protein or polypeptide)
         40164 CROWN
        529556 ETHER
       1173515 AMINO
<-------User Break----->
SEARCH ENDED BY USER
=> s (crown (w) ether) (P) ((amino(w) acid) or peptide or protein or
polypeptide)
         40164 CROWN
        529556 ETHER
       1173515 AMINO
       4600329 ACID
       396477 PEPTIDE
       2168943 PROTEIN
        109060 POLYPEPTIDE
L1
           451 (CROWN (W) ETHER) (P) ((AMINO(W) ACID) OR PEPTIDE OR PROTEIN OR
               POLYPEPTIDE)
=> s l1 (p) (benzoic or sulfonic or sulphonic)
         96713 BENZOIC
         84509 SULFONIC
          1603 SULPHONIC
             1 L1 (P) (BENZOIC OR SULFONIC OR SULPHONIC)
=> s 11 (p) (react or reaction or reacting or reacted or complex?)
        156460 REACT
       3183549 REACTION
        135032 REACTING
        198539 REACTED
       1829413 COMPLEX?
L3
           189 L1 (P) (REACT OR REACTION OR REACTING OR REACTED OR COMPLEX?)
=> s 13 and (BENZOIC OR SULFONIC OR SULPHONIC)
         96713 BENZOIC
         84509 SULFONIC
          1603 SULPHONIC
L4
             2 L3 AND (BENZOIC OR SULFONIC OR SULPHONIC)
=> d 14 bib ab 1-2
T.4
     ANSWER 1 OF 2 CAPLUS COPYRIGHT 2008 ACS on STN
    2005:36579 CAPLUS <<LOGINID::20080621>>
AN
DN
     142:114475
TΙ
     Chemical reagents capable of selective attachment to and reaction with
     peptides and proteins
IN
     Beauchamp, Jesse L.; Julian, Ryan R.; Stoltz, Brian M.; May, Jeremy A.
PA
SO
    U.S. Pat. Appl. Publ., 14 pp.
     CODEN: USXXCO
DT
   Patent
LA
   English
FAN.CNT 1
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	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 20050010059	A1	20050113	US 2004-782373	20040218
PRAI	US 2003-448290P	P	20030219		
AB	***Crown***	***ether	*** -conta.	biomimetic reagents,	e.a. (I) (R =

Q) and (II) (R = Q), capable of selectively forming non-covalent

complexes and initiating intermol. reactions with peptides in the gas phase are described and their reactions with primary amines and peptides in the gas phase was studied by electrospray ionization mass spectrometry (ESI-MS). The reagents are particularly useful in the synthesis of amine-contg. compds., and particularly gas phase

peptide chem. The invention also relates to the use of diazo-based reagents, e.g. (III) (R = R1 = Q; R= Q, R1 = Et), that bind to and become covalently attached to ***amino*** ***acid*** residues, particularly residues contg. primary amines. It further relates to the use of reagents contg. acidic groups or transition metal binding functionalities that initiate selective cleavage of ***amino***

acid residues, particularly residues contg. primary amines.

There is claimed a method of selectively forming noncovalent ***complexes***

and initiating intermol, reactions with amine-contg, compds, comprises ***reacting*** the amine-contg. compd. with a second compd. comprising at least one ***crown*** ***ether*** group and a moiety selected from acidic groups, transition metal binding groups and diazo groups, wherein the ***crown*** ***ether*** is 18-crown-6 ether and the acidic group is ***benzoic*** acid. Thus, 18-crown-6-methanol was treated with lithium diisopropylamine in THF at 70.degree. followed by etherification with 2,9-bis(bromomethyl)-1,10-phenanthroline in THF/CH2C12 at room temp. for 24 h gave the compd. I. ESI-MS of the compd. I, Cu(I), and H-Lvs-Lvs-OH (KK) indicated that compd. I formed an abundant noncovalent ***complex*** with the ***peptide*** KK and copper(I). Collisional activation of the base peak [1+KK+Cu+H]2+ resulted primarily in dissocn. of the ***complex*** into (1+Cu)+ and [KK+H]+ with an addnl. prominent peak corresponding to the loss of 44 Da from [KK+H]+. This loss is most likely explained as elimination of CO2 from the

complex [1+KK+Cu+2H]3+ vielded the loss of CO2 directly. In the absence of the Cu(I) ion, no loss of 44 Da was obsd. for either charge state, suggesting that Cu(I) effectively initiates this ***reaction***

C-terminus. Collisional activation of the much less abundant

L4 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2008 ACS on STN

1988:6386 CAPLUS << LOGINID:: 20080621>> AN

DN 108:6386

OREF 108:1219a,1222a

- The use of crown ethers in peptide chemistry. Part 1. Syntheses of amino acid complexes with the cyclic polyether 18-crown-6 and their oligomerization in dicyclohexylcarbodiimide-containing solutions
- AU Mascagni, Paolo; Hyde, Carolyn B.; Charalambous, Mario A.; Welham, Kevin
- CS Sch. Pharm., Univ. London, London, WC1N 1AX, UK
- SO Journal of the Chemical Society, Perkin Transactions 2: Physical Organic Chemistry (1972-1999) (1987), (3), 323-7 CODEN: JCPKBH: ISSN: 0300-9580
- DT Journal
- LA English
- O.S. CASREACT 108:6386

AB The synthesis of ***amino*** ***acid*** ***complexes*** with cyclic polyether 18-crown-6 and their soly, properties in org, solvents and dicyclohexylcarbodiimide as coupling agent. The mechanism leading to the formation of the oligopeptides involves transfer of one N-H proton from the ***crown*** ***ether*** ***complex*** to the carbodiimide nitrogen. => d his (FILE 'HOME' ENTERED AT 07:20:40 ON 21 JUN 2008) FILE 'CAPLUS' ENTERED AT 07:21:02 ON 21 JUN 2008 451 S (CROWN (W) ETHER) (P) ((AMINO(W) ACID) OR PEPTIDE OR PROTEIN T. 1 1.2 1 S L1 (P) (BENZOIC OR SULFONIC OR SULPHONIC) L3 189 S L1 (P) (REACT OR REACTION OR REACTING OR REACTED OR COMPLEX?) L42 S L3 AND (BENZOIC OR SULFONIC OR SULPHONIC) => S L3 AND (diazo) 35274 DIAZO 2 L3 AND (DIAZO) => s 15 not 14 1 L5 NOT L4 1.6 => d 16 bib ab ANSWER 1 OF 1 CAPLUS COPYRIGHT 2008 ACS on STN AN 2003:581554 CAPLUS <<LOGINID::20080621>> DN 140:321682 Biomimetic approaches to gas phase peptide chemistry: combining selective TI binding motifs with reactive carbene precursors to form molecular mousetraps AU Julian, Ryan R.; May, Jeremy A.; Stoltz, Brian M.; Beauchamp, J. L. CS Beckman Institute, California Institute of Technology, Pasadena, CA, 91125, USA SO. International Journal of Mass Spectrometry (2003), 228(2-3), 851-864 CODEN: IMSPF8: ISSN: 1387-3806

PB Elsevier Science B.V.

DT Journal

LA English

OS CASREACT 140:321682

CANEACT 140:321682
Biominetic reagents capable of selectively forming non-covalent complexes and initiating intermol. reactions with peptides in the gas phase are presented. In the present work, 18-crown-6 ether (18C6) is utilized to bind specifically to various protonated primary amines, including the protonated side chain of lysine. The use of multiple crown ethers is shown to be an efficient method for enhancing the binding energy, which is a crit. factor influencing the success of these reagents. The binding energy must exceed any reaction barriers to the desired chem., otherwise simple dissoon. of the complex occurs. Two reagents contg. acidic and transition metal binding functionalities, resp., designed to selectively cleave peptide bonds, are synthesized and tested exptl. A third class of reagent designed to covalently attach to peptides utilizing carbene insertion chem. is also presented. The results demonstrate that combining

the recognition and binding powers of 1805 with an easily activated ***diazo*** group allows for the efficient generation of a highly reactive carbene within a non-covalent complex. Intermol. insertion reactions initiated by the carbene can transform these non-covalent complexes into covalently bound mols. Electrospray ionization mass spectrometry and d. functional theory (DFT) are utilized to evaluate these intermol. insertion reactions. The results from expts. with several small mols. and peptides are presented. These ***diazo*** -based reagents prove to be highly versatile mols. capable of binding to, and with appropriate activation, becoming covalently attached to virtually any mol. that contains a primary amine. For this reason, they have been dubbed mol. moustraps.

RE.CNT 36 THERE ARE 36 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> d his

(FILE 'HOME' ENTERED AT 07:20:40 ON 21 JUN 2008)

=> s 13 and polyamine

37008 POLYAMINE

L7 0 L3 AND POLYAMINE

1 S L5 NOT L4

=> d his

L6

(FILE 'HOME' ENTERED AT 07:20:40 ON 21 JUN 2008)

L4 2 S L3 AND (BENZOIC OR SULFONIC OR SULPHONIC)
L5 2 S L3 AND (DIAZO)

L6 1 S L5 NOT L4 L7 0 S L3 AND POLYAMINE

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